

21	22	23	24	25	Σ

Worksheet No. 5 Advanced Mathematics II

Exercise 21: Let

$$C = \begin{pmatrix} 5 & 4 & -5 & 5 \\ -1 & 1 & 2 & -2 \\ -3 & -4 & 7 & -5 \\ -4 & -5 & 7 & -5 \end{pmatrix}.$$

Calculate all eigenvalues and eigenvectors of C .

Exercise 22: Given two planes $E : 2x_1 - x_2 - 2x_3 = 0$ and $F : x(\lambda, \mu) = (0, 1, 0)^\top + \lambda(4, 1, 3)^\top + \mu(4, -7, 3)^\top$, $\lambda, \mu \in \mathbb{R}$. Let the linear mapping $\Phi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ be a reflection in E and the mapping $\Psi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ the orthogonal projection on F .

- Determine the matrices A of Φ and B of Ψ with respect to the standard basis. Check if A or B are orthogonal matrices and compute the matrix products A^2 as well as B^2 .
- Compute the eigenvalues and eigenvectors of Φ and Ψ .

Exercise 23: The linear mappings $\Phi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ and $\Psi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ are represented by the following matrices with respect to the standard basis:

$$A = \frac{1}{3} \begin{pmatrix} 2 & -1 & 1 \\ -1 & 2 & 1 \\ 1 & 1 & 2 \end{pmatrix}, \quad B = \frac{1}{9} \begin{pmatrix} 7 & 4 & -4 \\ 4 & 1 & 8 \\ -4 & 8 & 1 \end{pmatrix}.$$

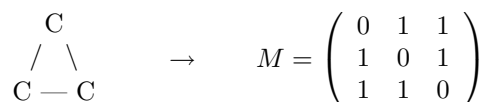
- Compute the eigenvalues and eigenvectors of A and B .
- Check if Φ and Ψ describe a projection, reflection or rotation. If so, find the projection plane, reflection plane or rotation axis and angle of rotation.

Exercise 24:

- Determine the eigenvalues of $A = \begin{pmatrix} 5 & 0 & 4 \\ 0 & -6 & 0 \\ 1 & 0 & 2 \end{pmatrix}$.
- Let $p(x) = c_3x^3 + c_2x^2 + c_1x + c_0$ be the characteristic polynomial of A . Show that $p(A) = 0$, i.e. $c_3A^3 + c_2A^2 + c_1A + c_0I_3 = 0$. (This is true in general for every square matrix and its characteristic polynomial!)
- From $p(A) = 0$, determine the inverse matrix A^{-1} .

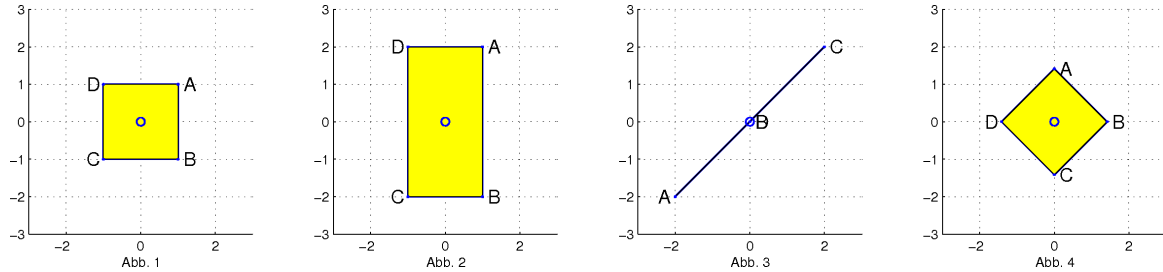
Exercise 25: *In physical chemistry the molecular orbitals and the corresponding energies of π -electron systems can be computed quantum mechanically via the Hückel method. The molecular structure is represented in a kind of normed adjacency matrix, where every carbon compound is represented by a 1. The eigenvalues of this matrix characterise the energies, and the eigenvectors describe the structure of the molecular orbitals.*

Compute the eigenvalues and eigenvectors of the matrix M for the illustrated cyclopropenyl compound:



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Exercise T17: Figure 1 shows a square around the origin with the vertices A, B, C, D . The following figures show the images of the square obtained by applying different linear mappings $\Phi_{1,2,3} : \mathbb{R}^2 \rightarrow \mathbb{R}^2$:



Determine the eigenvalues of the linear mappings. Sketch the eigenvectors if possible.

Exercise T18:

- (a) Compute the characteristic polynomial $p(\lambda) = \det(A - \lambda I)$ of A for $\lambda \in \mathbb{R}$, where

$$I = \begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{pmatrix}, \quad A = \begin{pmatrix} 2 & 1 & 2 \\ 1 & 2 & 2 \\ 1 & 1 & 3 \end{pmatrix}.$$

- (b) Determine the eigenvalues, i.e. zeros $\lambda_1, \lambda_2, \lambda_3$ of the polynomial $p(\lambda)$.
 (c) Find the eigenvectors, i.e. the solution set of the homogeneous system of linear equation $(A - \lambda_j I)x = 0$ for $j = 1, 2, 3$.

Exercise T19: Let the linear mapping $\Phi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ be a rotation around the x_2 -axis by the angle $\frac{\pi}{3}$ and let the linear mapping $\Psi : \mathbb{R}^3 \rightarrow \mathbb{R}^3$ be a reflection on the plane $E : x_1 + 2x_2 = 0$.

- (a) Determine the matrices A of Φ and B of Ψ with respect to the standard basis. Check if A and B are orthogonal matrices.
 (b) Compute the eigenvalues of A and B , as well as the eigenvectors to the real eigenvalues.

Exercise T20: Determine all eigenvalues of the square matrices

$$(a) \quad A = \begin{pmatrix} 2 & 7 & -2 \\ -1 & -1 & -1 \\ -1 & -5 & 3 \end{pmatrix}, \quad (b) \quad B = \frac{1}{2} \begin{pmatrix} 1 & -5 & 5 \\ 5 & -9 & 5 \\ 5 & -5 & 1 \end{pmatrix},$$

and an eigenvector to every real eigenvalue λ .